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On-Surface Synthesis and Characterization of [7]Triangulene Quantum Ring

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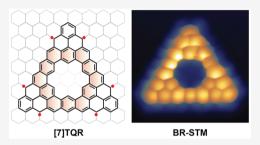
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ABSTRACT: The ability to engineer geometrically well-defined antidots in large triangulene homologues allows for creating an entire family of triangulene quantum rings (TQRs) with tunable high-spin ground state, crucial for next-generation molecular spintronic devices. Herein, we report the synthesis of an open-shell [7] triangulene quantum ring ([7] TQR) molecule on Au(111) through the surface-assisted cyclodehydrogenation of a rationally designed kekulene derivative. Bond-resolved scanning tunneling microscopy (BR-STM) unambiguously imaged the molecular backbone of a single [7] TQR with a triangular zigzag edge topology, which can be viewed as [7] triangulene decorated with a coronene-like antidot in the center. Additionally, dI/dV mapping reveals that both inner and



outer zigzag edges contribute to the edge-localized and spin-polarized electronic states of [7]TQR. Both experimental results and spin-polarized density functional theory calculations indicate that [7]TQR retains its open-shell septuple ground state (S = 3) on Au(111). This work demonstrates a new route for the design of high-spin graphene quantum rings for future quantum devices.

KEYWORDS: Triangulene quantum ring, antidot engineering, on-surface synthesis, open-shell, scanning probe microscopy

pen-shell graphene nanostructures (GNs) have attracted tremendous interest due to their unique electronic and magnetic properties, which make them promising candidates for carbon-based molecular spintronics. 1-4 The magnetic ordering of these open-shell GNs can be precisely tailored via engineering their topological structures and edge configurations. 5,6 In particular, cutting triangular motifs along the zigzag orientation of graphene creates an entire family of triangulenes with high-spin ground states and large spin quantum number scaling linearly with the number of carbon atoms at their zigzag edges.^{7–9} The ground state spin quantum number of the triangulenes is associated with the sublattice imbalance, as predicted by Ovchinnikov's rule and Lieb's theorem for bipartite lattices. ^{10,11} As such, the ground state spin quantum number (S) of triangulenes is equal to $\frac{|N_A - N_B|}{2}$, where N_A and N_B represent the number of carbon atoms from the two interpenetrating sublattices (A and B), respectively.

Recently, [3]-, [4]-, and [5]triangulene molecules have been successfully fabricated on different surfaces and their chemical structures and spin-polarized edge states have been characterized by scanning probe techniques. 12-14 A further extension of triangulene molecules to a larger system not only increases their ground state spin quantum number but also offers higher feasibility for device fabrication. Apart from tuning the size of triangulenes, topological engineering via creating an antidot (denoted as antidot engineering) in large triangulene homologues provides an alternative approach to fabricate a series of intriguing triangulene quantum rings (TQRs) with

tailored spin quantum number and magnetic ordering associated with both inner and outer zigzag edges. ^{16–19} Compared to the triangulene counterparts, the presence of an antidot in TQRs results in a larger number of zigzag edges per unit area. Moreover, the spin-polarized energy gap and spin density distribution of TQRs, together with the magnetic ordering in both inner and outer zigzag edges, can be precisely controlled via tailoring the topology of antidots. ^{20–22} Therefore, antidot engineering constitutes a flexible and versatile strategy toward the fabrication of a plethora of novel magnetic quantum nanostructures, crucial to unlocking their full potential in next-generation spintronics applications.

Analogously to triangulenes, the ground state spin quantum number of TQRs can also be predicted by Ovchinnikov's rule by evaluating their sublattice imbalance. As illustrated in Figure 1a, excavating a coronene moiety from the center of an intact [7] triangulene molecule (Figure 1a, right) generates a [7] TQR molecule containing a hexagonal coronene-like antidot (Figure 1a, middle right). Notably, the as-removed hexagonal motif consists of an equal number of two sublattices (A- and B-type). As a result, the ground state spin quantum number (S = 3) is

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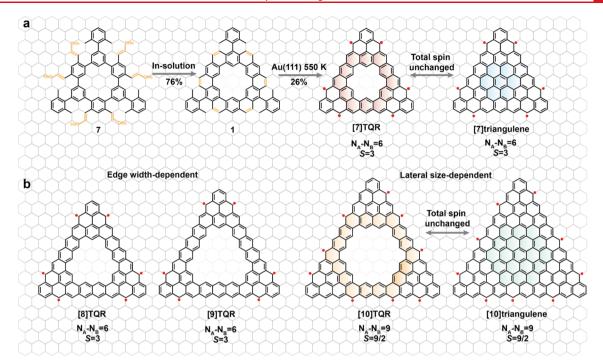


Figure 1. Schematic illustration of the synthesis of [7]TQR via antidot engineering of triangulene molecules. (a) Illustration of the synthetic route toward [7]TQR. [7]TQR with a coronene-like antidot and [7]triangulene have the same ground state spin quantum number of S=3. (b) Engineering of triangular and hexagonal antidots within π -extended triangulene molecules. Incorporating a triangular antidot in the center of triangulene modifies the molecules' ground state spin quantum number. In contrast, introducing a hexagonal antidot in both [7]- and [10]triangulene retains the ground state spin quantum number.

preserved after introducing such a hexagonal antidot in [7]triangulene. In contrast to [7]triangulene, the presence of inner zigzag edges in [7]TQR is expected to modulate its spin density and magnetic ordering in this system.²¹ Similarly, larger TQRs also retain the same high spin quantum number after removing hexagonal antidots (e.g., high-order coronene-like moieties) in the triangulene hosts (e.g., [10]TQR, Figure 1b, right). By contrast, incorporating triangular antidots into triangulenes can reduce the sublattice imbalance and thus lower their ground-state spin quantum numbers, as exemplified in the case of [8]TQR and [9]TQR (Figure 1b, left). Both of them have the same ground state spin quantum number of S =3, lower than that of [8] triangulene $(S = \frac{7}{2})$ and [9]triangulene (S = 4). In this case, the ground state spin quantum number S only depends on the width of the ring, as defined by the number of benzene units (N_W) between outer and inner edges: $S = 3(N_W + 1)^{21}$ Therefore, the ground state spin quantum number of TQRs can be modulated by tailoring the topology of the antidots, including the lateral size (Figure 1b, right) and width of the ring (Figure 1b, left).

To demonstrate the concept of antidot engineering, we designed a new precursor 1 containing a kekulene core attached by three 2,6-dimethylphenyl substituents at mesopositions of three alternating edges. The kekulene is well-known for its annelated macrocyclic topology that provides an ideal platform for antidot engineering. The dehydrogenation of methyl groups of precursor 1 generates benzylic radicals, which then couple with the neighboring carbon atoms of the kekulene core. Subsequently, an aromaticity-driven ring-closure reaction leads to the formation of [7]TQR (Figure 1a). Precursor 1 was obtained via multiple solution-phase synthetic steps as inspired by our previous work (see details in the Supporting Information).²⁸ In brief, a macrocyclic oligo(m-

phenylene) intermediate 7 carrying six vinyl ether groups was first synthesized by Suzuki coupling reaction. The subsequent ${\rm Bi}({\rm OTf})_3$ catalyzed cyclization reaction of 7 in 1,2-dichloroethane at 90 °C led to the formation of the 2,6-dimethylphenyl-substituted kekulene 1 in 76% yield. The structure of the precursor 1 was confirmed by $^1{\rm H}$ NMR, $^{13}{\rm C}$ NMR spectra, and mass spectra.

Precursor 1 was first deposited onto Au(111) at submonolayer coverage under ultrahigh vacuum conditions. It exhibits a three-lobed appearance in the STM image attributed to its 3-fold structural symmetry (Figure S1). Subsequently, postannealing of Au(111) substrate at 550 K for 10 min triggers the surface-assisted cyclodehydrogenation of 1 toward the formation of [7]TQR.²⁹ An overview STM image captures that as-formed products contain both isolated triangular-shaped molecules and irregular oligomers presumably arising from intermolecular coupling on Au(111) during the thermal annealing process (Figure 2a, Figure S1). Isolated triangular-shaped molecules constitute approximately 26% of all the products, as estimated by a statistical analysis of multiple STM images. In addition, a magnified STM image of triangular-shaped products with a metallic tip resolves a dim hole-like feature in the molecular center (Figure 2b and Figure S1) and characteristic edge-localized nodal patterns, resembling the patterns observed at the zigzag edges of the zigzagedged graphene nanostructures in the previous reports. ^{13,14,29,30} To probe the chemical structure of an isolated product, we performed bond-resolved STM (BR-STM) imaging with a carbon monoxide-functionalized tip (CO-tip). The BR-STM imaging was conducted at constantheight mode with reduced tip-sample distances (i.e., Pauli repulsion regime), wherein the CO molecule undergoes a lateral relaxation over the areas with high electron density

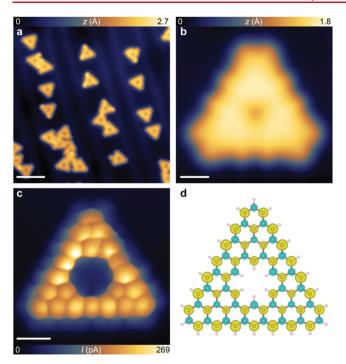


Figure 2. Structural characterization of [7]TQR synthesized on Au(111). (a) Large-scale STM image of [7]TQR molecules on Au(111) after annealing the precursor 1 at 550 K (V=1.0~V, I=300~PA). (b) Magnified high-resolution STM image of an individual [7]TQR molecule (V=-0.61~V, I=600~PA). (c) Corresponding BR-STM image ($V=5~mV, \Delta z=-1.1~Å$; set point prior to turn off feedback, V=50~mV, I=300~pA). (d) DFT-calculated spin density distribution of a [7]TQR molecule (DMRG-calculated spin density, which agrees well with the DFT one, is shown in Figure S8b). Scale bar in panel a = 4 nm, and in panels b and c = 5 Å.

(chemical bonds), which modulates the overall tunneling conductance between tip and sample, resulting in sharp features associated with the chemical bonds in the tunneling current image. The corresponding BR-STM image of a single triangular-shaped molecule (Figure 2c) unambiguously resolves the molecular backbone consisting of 12 fused benzene rings with a clear-cut hexagonal antidot (a diameter of 5.7 Å) in the molecular center. Furthermore, the flat adsorption geometry of final products on Au(111) indicates the absence of any chemical bonding of molecule to the surface. These observations confirm the successful synthesis of the [7]TQR on Au(111).

To unveil the electronic structure of [7]TQR, we acquired differential conductance spectra (dI/dV) over a single molecule on Au(111) using a calibrated metallic tip. The characteristic point dI/dV spectra collected at the corner (red curve) and the edge center (blue curve) of a [7]TQR, along with the reference spectrum recorded on bare Au(111) (gray dashed curve), are presented in Figure 3a. The dI/dV spectrum collected at the edge corner of the molecule (red cross in the inset image in Figure 3a) shows two pronounced peaks at $-0.61 \text{ V } (\pm 0.05 \text{ V}) \text{ and } +0.91 \text{ V } (\pm 0.05 \text{ V}), \text{ presumably}$ corresponding to the positive and negative ion resonances (PIR and NIR), respectively. We also observed a weak "bump" feature around -0.3 V (± 0.05 V) that overlaps with the Au(111) surface states, which can be better resolved in dI/dVspectrum acquired in a smaller bias range (Figure S3). In addition, a prominent peak at -1.47 V (± 0.05 V) and a substantial rise above +2.0 V can be observed in dI/dV spectra acquired at both the corner and the edge center.

In order to gain a more in-depth insight into the electronic structures of [7]TQR, we performed spin-polarized density functional theory (DFT) calculations using hybrid PBE0 XC

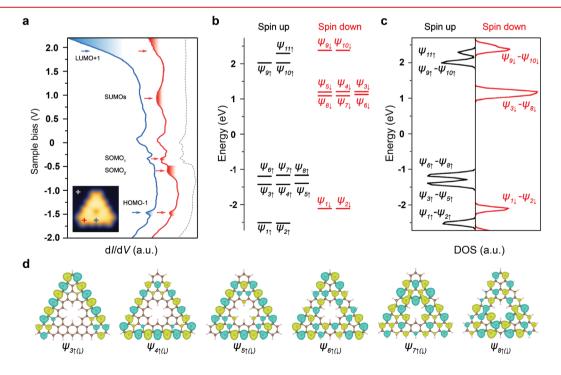


Figure 3. Electronic structure characterization of [7] TQR. (a) Point dI/dV spectra acquired over different sites of [7] TQR and Au(111) substrate. dI/dV curves taken at the corner (red curve), center of the edge (blue curve), and taken on Au(111) (gray-dashed curve). (b) DFT-calculated spin-polarized molecular orbital energy diagram of a freestanding [7] TQR. The black and red symbols refer to spin-up and spin-down states, respectively. (c) Calculated density of states (DOS) of a freestanding [7] TQR. (d) DFT calculated wave functions of six pairs of SOMOs and SUMOs (ψ_{311} – ψ_{811}). The blue and green isosurface colors indicate opposite phases of the wave function. (a.u., arbitrary units).

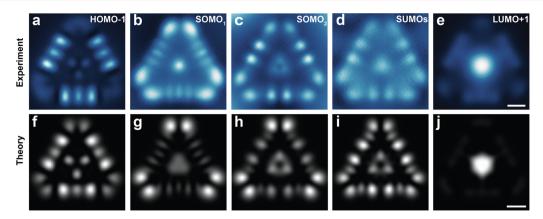


Figure 4. Experimental and calculated dI/dV maps of [7]TQR. (a-e) Constant-current dI/dV maps recorded at different energy positions. Panels a and e refer to the HOMO -1 (-1.47 V) and LUMO +1 (+2.20 V) of [7]TQR, respectively. Panels b-d refer to the SOMO₁ (-0.61 V), SOMO₂ (-0.3 V), and SUMOs (+0.91 V) of [7]TQR, respectively. (f-j) Simulated dI/dV maps acquired at different energy positions corresponding to different sets of orbitals: (f) ψ_{11} - ψ_{21} , (g) ψ_{31} - ψ_{51} , (h) ψ_{61} - ψ_{81} , (i) ψ_{31} - ψ_{81} , and (j) ψ_{91} - ψ_{101} . Scale bar: 5 Å.

(exchange-correlation) functional^{37,38} and density matrix renormalization group (DMRG) calculations employing the Hubbard model Hamiltonian of free-standing molecule. 39,40 Both methods reveal that the septuple state (S = 3) of freestanding [7]TQR is the ground state. Our DFT calculation of free-standing [7] TQR predicts that the ground state S = 3 is lower in energy by 378 and 448 meV than its S = 1 open-shell states, respectively (Figure 2d, Figure S2), consistent with the theoretical predictions of the triangulene systems. 13,14 In addition, we also carried out fully relaxed DFT calculations of [7]TQR molecule on Au(111) using PBE XC-functional, which confirms that its S = 3 ground state remains stable on a metallic substrate. 41 The total energy DFT calculations reveal that the molecule is only weakly coupled to the surface mainly by dispersion interactions at a position of ~ 3.3 Å above the surface plane. The calculated spin density of the molecule on the surface (Figure S6) shows very similar character to that of the free-standing molecule as shown in Figure 2d. In addition, we performed DMRG calculations with the reduced on-site repulsion to mimic the effect of metallic substrate on the electronic structure of the adsorbed [7]TQR molecule.42 According to these DMRG calculations, the S = 3 ground state remains stable even when a half on-site repulsion energy is used (refer to Figure S8 for more details).

Figure 3b presents the calculated single-particle DFT spinpolarized energy level diagram of a freestanding [7]TQR in the ferromagnetic septuple state (S = 3). The frontier orbitals of a neutral [7]TQR (Figure 3a) contain six singly occupied molecular orbitals (denoted as SOMOs including spin-up $\psi_{3\uparrow} - \psi_{8\uparrow}$) and six singly unoccupied molecular orbitals (denoted as SUMOs including spin-down $\psi_{3\downarrow}$ – $\psi_{8\downarrow}$) with the corresponding wave function plots shown in Figure 3d. It is noted that a large energy splitting of 200 meV between $\psi_{3\uparrow} - \psi_{5\uparrow}$ and $\psi_{6\uparrow} - \psi_{8\uparrow}$ of SOMOs leads to a double peak feature associated with SOMOs in density of states (DOS) (Figure 3c). This is consistent with our experimental observations of a pair of occupied electronic states at −0.61 and -0.3 V, which thus can be assigned as SOMO1 and SOMO₂ (Figure 3c), respectively. In contrast to SOMOs, a reduced energy splitting between SUMOs leads to a single peak feature of SUMO-derived DOS, in agreement with the experimental observation of a single broad peak located at +0.91 V (labeled as SUMOs in Figure 3c). Such a discrepancy between SOMO- and SUMO-derived DOS features can be

attributed to their different degrees of orbital degeneracy (Figure 3b). According to the energy ordering of these molecular orbitals, a sharp peak at -1.47 V (± 0.05 V) and a substantial rise above +2.0 V can be assigned to HOMO -1 (highest occupied molecular orbital) and LUMO + 1 (lowest unoccupied molecular orbital) derived from $\psi_{1\downarrow}-\psi_{2\downarrow}$ and $\psi_{9\uparrow}-\psi_{10\uparrow}$, respectively.

To resolve the spatial distribution of these molecular orbitals, we performed the differential conductance mapping at the corresponding energetic positions of the aforementioned peaks (Figure 4a-e). dI/dV maps collected at -0.61 V (Figure 4b), -0.3 V, and +0.91 V (Figure 4d) reveal the edge localization and the characteristic nodal patterns for both the outer and inner edges of [7]TQR, wherein the nodal patterns of outer edges resemble that of π -extended triangulenes in previous reports. 13,14 As for the triangulenes, SOMO- and SUMO-derived states often show identical spatial patterns. However, it is noted that dI/dV maps acquired at -0.61~V $(SOMO_1)$, $-0.30 \text{ V} (SOMO_2)$, and +0.91 V (SUMOs) exhibit slightly different spatial patterns. Such a difference can be attributed to the different frontier orbitals comprising SOMO₁, SOMO₂, and SUMOs, respectively. Notably, SOMOs with a lower degree of degeneracy split into two subsets of nearly degenerate orbitals $(\psi_{3\uparrow} - \psi_{5\uparrow}, SOMO_1, and \psi_{6\uparrow} - \psi_{8\uparrow}, SOMO_2)$ (Figure 3b,c), resulting in slightly different patterns in the dI/ dV maps. All the key features of experimental dI/dV maps taken at -0.61 V and -0.3 V (Figure 4c) are well reproduced in the calculated dI/dV maps involving the orbitals of $\psi_{3\uparrow}$ – $\psi_{5\uparrow}$ (SOMO₁) and $\psi_{6\uparrow} - \psi_{8\uparrow}$ (SOMO₂).⁴³ In contrast to SOMOs, SUMOs of [7]TQR with a higher degree of degeneracy (Figure 3b) yield a single peak in dI/dV spectrum. A direct comparison between experimental and theoretical dI/dV map reveals that orbitals $\psi_{5\downarrow}$ and $\psi_{6\downarrow}$ with similar wave function patterns (Figure 3d) have a considerable contribution (high weightage) to the dI/dV map of SUMOs. Such observations further imply that these probed orbitals, indeed, correspond to the frontier SOMOs and SUMOs of [7]TQR.

In addition, dI/dV map recorded at -1.47 V (HOMO -1) reveals a triangular lobe at the center and characteristic nodal pattern (three bright lobes) at the outer edges. In contrast, dI/dV map taken at +2.20 V (LUMO +1) exhibits a hexagonal protrusion at the center and faint feature at the outer edges, suggesting a larger contribution from the inner edge states to LUMO +1. Moreover, the calculated dI/dV images involving

the spin-down $\psi_{1\downarrow}-\psi_{2\downarrow}$ and spin-up $\psi_{9\uparrow}-\psi_{10\uparrow}$ orbitals show excellent agreement with d*I*/d*V* maps taken at -1.47 V (HOMO - 1) and +2.20 V (LUMO + 1), respectively (Figure 4f,j).

Finally, the energy gap of Au-supported [7]TQR is determined to be 1.21 eV based on the energy separation between SOMO₂ and SUMOs (Figure 3A), lower than the theoretically predicted value (2.25 eV) of a free-standing [7]TQR. This can be attributed to the gap reduction arising from the screening effect from the metallic Au(111) surface, consistent with the observation in π -extended triangulenes and other similar systems. ^{13,14,42,44–47} All these observations discussed above manifest that the septuple magnetic ground state (S = 3) of [7]TQR is likely to be retained on Au(111).

In summary, we have demonstrated a new synthetic route to use the substituted kekulene as the precursor for the synthesis of unsubstituted [7]TQR with a coronene-like antidot on Au(111). BR-STM characterization unambiguously reveals the chemical structure of [7]TQR at the submolecular level, while dI/dV measurements unveil the spin-polarized electronic states contributed from both inner and outer zigzag edges of [7]TQR. Both experimental data and theoretical calculations suggest that [7]TQR retains its open-shell septuple ground state (S=3) on Au(111). The successful synthesis of a triangulene quantum ring via antidot engineering opens up a new avenue for the fabrication of high-spin quantum nanostructures toward next-generation quantum devices.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.nanolett.0c04627.

Details of experimental methods, supporting experimental and theoretical data, synthetic procedures, NMR characterizations and high-resolution mass spectra (PDF)

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The authors declare no competing financial interest.

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REFERENCES

- (1) Morita, Y.; Suzuki, S.; Sato, K.; Takui, T. Synthetic Organic Spin Chemistry for Structurally Well-Defined Open-Shell Graphene Fragments. *Nat. Chem.* **2011**, *3* (3), 197–204.
- (2) Han, W.; Kawakami, R. K.; Gmitra, M.; Fabian, J. Graphene Spintronics. *Nat. Nanotechnol.* **2014**, *9* (10), 794–807.
- (3) Sun, Z.; Wu, J. Open-Shell Polycyclic Aromatic Hydrocarbons. *J. Mater. Chem.* **2012**, 22 (10), 4151–4160.
- (4) Wang, W. L.; Meng, S.; Kaxiras, E. Graphene NanoFlakes with Large Spin. *Nano Lett.* **2008**, 8 (1), 241–245.
- (5) Yazyev, O. V. Emergence of Magnetism in Graphene Materials and Nanostructures. *Rep. Prog. Phys.* 2010, 73 (5), 056501.
 (6) Fujii, S.; Enoki, T. Nanographene and Graphene Edges:
- (6) Fujii, S.; Enoki, T. Nanographene and Graphene Edges: Electronic Structure and Nanofabrication. *Acc. Chem. Res.* **2013**, 46 (10), 2202–2210.
- (7) Fernández-Rossier, J.; Palacios, J. J. Magnetism in Graphene Nanoislands. *Phys. Rev. Lett.* **2007**, 99 (17), 177204.
- (8) Güçlü, A. D.; Potasz, P.; Hawrylak, P. Zero-Energy States of Graphene Triangular Quantum Dots in a Magnetic Field. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, 88 (15), 155429.

- (9) Wang, W. L.; Yazyev, O. V.; Meng, S.; Kaxiras, E. Topological Frustration in Graphene Nanoflakes: Magnetic Order and Spin Logic Devices. *Phys. Rev. Lett.* **2009**, *102* (15), 157201.
- (10) Ovchinnikov, A. A. Multiplicity of the Ground State of Large Alternant Organic Molecules with Conjugated Bonds. *Theoret. Chim. Acta* **1978**, 47 (4), 297–304.
- (11) Lieb, E. Two Theorems on the Hubbard Model. *Phys. Rev. Lett.* **1989**, *62* (10), 1201–1204.
- (12) Pavliček, N.; Mistry, A.; Majzik, Z.; Moll, N.; Meyer, G.; Fox, D. J.; Gross, L. Synthesis and Characterization of Triangulene. *Nat. Nanotechnol.* **2017**, *12* (4), 308–311.
- (13) Mishra, S.; Beyer, D.; Eimre, K.; Liu, J.; Berger, R.; Gröning, O.; Pignedoli, C. A.; Müllen, K.; Fasel, R.; Feng, X.; Ruffieux, P. Synthesis and Characterization of Π-Extended Triangulene. *J. Am. Chem. Soc.* **2019**, *141* (27), 10621–10625.
- (14) Su, J.; Telychko, M.; Hu, P.; Macam, G.; Mutombo, P.; Zhang, H.; Bao, Y.; Cheng, F.; Huang, Z.-Q.; Qiu, Z.; Tan, S. J. R.; Lin, H.; Jelínek, P.; Chuang, F.-C.; Wu, J.; Lu, J. Atomically Precise Bottom-Up Synthesis of Π-Extended [5] Triangulene. *Sci. Adv.* **2019**, *5* (7), eaav7717.
- (15) Su, J.; Telychko, M.; Song, S.; Lu, J. Triangulenes: From Precursor Design to on-Surface Synthesis and Characterization. *Angew. Chem., Int. Ed.* **2020**, *59*, 7658–7668.
- (16) Yang, H.-X.; Chshiev, M.; Boukhvalov, D. W.; Waintal, X.; Roche, S. Inducing and Optimizing Magnetism in Graphene Nanomeshes. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2011**, 84 (21), 214404.
- (17) Pedersen, T. G.; Flindt, C.; Pedersen, J.; Mortensen, N. A.; Jauho, A.-P.; Pedersen, K. Graphene Antidot Lattices: Designed Defects and Spin Qubits. *Phys. Rev. Lett.* **2008**, *100* (13), 136804.
- (18) Pawlak, R.; Liu, X.; Ninova, S.; D'Astolfo, P.; Drechsel, C.; Sangtarash, S.; Häner, R.; Decurtins, S.; Sadeghi, H.; Lambert, C. J.; Aschauer, U.; Liu, S.-X.; Meyer, E. Bottom-Up Synthesis of Nitrogen-Doped Porous Graphene Nanoribbons. *J. Am. Chem. Soc.* **2020**, *142* (29), 12568–12573.
- (19) Moreno, C.; Vilas-Varela, M.; Kretz, B.; García-Lekue, A.; Costache, M. V.; Paradinas, M.; Panighel, M.; Ceballos, G.; Valenzuela, S. O.; Peña, D.; Mugarza, A. Bottom-Up Synthesis of Multifunctional Nanoporous Graphene. *Science* **2018**, *360* (6385), 199–203.
- (20) Bahamon, D. A.; Pereira, A. L. C.; Schulz, P. A. Inner and Outer Edge States in Graphene Rings: a Numerical Investigation. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2009**, 79 (12), 125414.
- (21) Potasz, P.; Güçlü, A. D.; Voznyy, O.; Folk, J. A.; Hawrylak, P. Electronic and Magnetic Properties of Triangular Graphene Quantum Rings. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2011**, 83 (17), 174441.
- (22) Grujić, M.; Tadić, M.; Peeters, F. M. Antiferromagnetism in Hexagonal Graphene Structures: Rings Versus Dots. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, 87 (8), 085434.
- (23) Staab, H. A.; Diederich, F. Cycloarenes, a New Class of Aromatic Compounds, I. Synthesis of Kekulene. *Chem. Ber.* **1983**, *116* (10), 3487–3503.
- (24) Buttrick, J. C.; King, B. T. Kekulenes, Cycloarenes, and Heterocycloarenes: Addressing Electronic Structure and Aromaticity Through Experiments and Calculations. *Chem. Soc. Rev.* **2017**, *46* (1), 7–20.
- (25) Pozo, I.; Majzik, Z.; Pavliček, N.; Melle-Franco, M.; Guitián, E.; Peña, D.; Gross, L.; Pérez, D. Revisiting Kekulene: Synthesis and Single-Molecule Imaging. *J. Am. Chem. Soc.* **2019**, *141* (39), 15488–15493.
- (26) Di Giovannantonio, M.; Yao, X.; Eimre, K.; Urgel, J. I.; Ruffieux, P.; Pignedoli, C. A.; Müllen, K.; Fasel, R.; Narita, A. Large-Cavity Coronoids with Different Inner and Outer Edge Structures. *J. Am. Chem. Soc.* **2020**, *142* (28), 12046–12050.
- (27) Fan, Q.; Martin-Jimenez, D.; Werner, S.; Ebeling, D.; Koehler, T.; Vollgraff, T.; Sundermeyer, J.; Hieringer, W.; Schirmeisen, A.; Gottfried, J. M. On-Surface Synthesis and Characterization of a

- Cycloarene: C108 Graphene Ring. J. Am. Chem. Soc. 2020, 142 (2), 894–899.
- (28) Fan, W.; Han, Y.; Dong, S.; Li, G.; Lu, X.; Wu, J. Facile Synthesis of Aryl-Substituted Cycloarenes via Bismuth(III) Triflate-Catalyzed Cyclization of Vinyl Ethers. CCS Chem. 2020, 1445–1452.
- (29) Ruffieux, P.; Wang, S.; Yang, B.; Sánchez-Sánchez, C.; Liu, J.; Dienel, T.; Talirz, L.; Shinde, P.; Pignedoli, C. A.; Passerone, D.; Dumslaff, T.; Feng, X.; Müllen, K.; Fasel, R. On-Surface Synthesis of Graphene Nanoribbons with Zigzag Edge Topology. *Nature* **2016**, 531 (7595), 489–492.
- (30) Talirz, L.; Söde, H.; Cai, J.; Ruffieux, P.; Blankenburg, S.; Jafaar, R.; Berger, R.; Feng, X.; Müllen, K.; Passerone, D.; Fasel, R.; Pignedoli, C. A. Termini of Bottom-Up Fabricated Graphene Nanoribbons. J. Am. Chem. Soc. 2013, 135 (6), 2060–2063.
- (31) Nguyen, G. D.; Tsai, H.-Z.; Omrani, A. A.; Marangoni, T.; Wu, M.; Rizzo, D. J.; Rodgers, G. F.; Cloke, R. R.; Durr, R. A.; Sakai, Y.; Liou, F.; Aikawa, A. S.; Chelikowsky, J. R.; Louie, S. G.; Fischer, F. R.; Crommie, M. F. Atomically Precise Graphene Nanoribbon Heterojunctions From a Single Molecular Precursor. *Nat. Nanotechnol.* **2017**, *12* (11), 1077–1082.
- (32) Li, J.; Merino-Díez, N.; Carbonell-Sanromà, E.; Vilas-Varela, M.; de Oteyza, D. G.; Peña, D.; Corso, M.; Pascual, J. I. Survival of Spin State in Magnetic Porphyrins Contacted by Graphene Nanoribbons. *Sci. Adv.* **2018**, *4* (2), eaaq0582.
- (33) Li, J.; Sanz, S.; Castro-Esteban, J.; Vilas-Varela, M.; Friedrich, N.; Frederiksen, T.; Peña, D.; Pascual, J. I. Uncovering the Triplet Ground State of Triangular Graphene Nanoflakes Engineered with Atomic Precision on a Metal Surface. *Phys. Rev. Lett.* **2020**, *124* (17), 177201.
- (34) Song, S.; Guo, N.; Li, X.; Li, G.; Haketa, Y.; Telychko, M.; Su, J.; Lyu, P.; Qiu, Z.; Fang, H.; Peng, X.; Li, J.; Wu, X.; Li, Y.; Su, C.; Koh, M. J.; Wu, J.; Maeda, H.; Zhang, C.; Lu, J. Real-Space Imaging of a Single-Molecule Monoradical Reaction. *J. Am. Chem. Soc.* **2020**, *142* (31), 13550–13557.
- (35) Hapala, P.; Kichin, G.; Wagner, C.; Tautz, F. S.; Temirov, R.; Jelínek, P. Mechanism of High-Resolution STM/AFM Imaging with Functionalized Tips. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2014**, 90 (8), 085421.
- (36) Hapala, P.; Temirov, R.; Tautz, F. S.; Jelínek, P. Origin of High-Resolution IETS-STM Images of Organic Molecules with Functionalized Tips. *Phys. Rev. Lett.* **2014**, *113* (22), 226101.
- (37) Blum, V.; Gehrke, R.; Hanke, F.; Havu, P.; Havu, V.; Ren, X.; Reuter, K.; Scheffler, M. Ab Initio Molecular Simulations with Numeric Atom-Centered Orbitals. *Comput. Phys. Commun.* **2009**, *180* (11), 2175–2196.
- (38) Perdew, J. P.; Ernzerhof, M.; Burke, K. Rationale for Mixing Exact Exchange with Density Functional Approximations. *J. Chem. Phys.* **1996**, *105* (22), 9982–9985.
- (39) Schollwöck, U. The Density-Matrix Renormalization Group. *Rev. Mod. Phys.* **2005**, *77* (1), 259–315.
- (40) Brabec, J.; Brandejs, J.; Kowalski, K.; Xantheas, S.; Legeza, Ö.; Veis, L. Massively Parallel Quantum Chemical Density Matrix Renormalization Group Method. *arXiv* 2020, arXiv:2001.04890v2. https://arxiv.org/abs/2001.04890 (June 19th, 2020).
- (41) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. *Phys. Rev. Lett.* **1996**, 77 (18), 3865–3868.
- (42) Neaton, J. B.; Hybertsen, M. S.; Louie, S. G. Renormalization of Molecular Electronic Levels at Metal-Molecule Interfaces. *Phys. Rev. Lett.* **2006**, *97* (21), 216405.
- (43) Krejčí, O.; Hapala, P.; Ondráček, M.; Jelínek, P. Principles and Simulations of High-Resolution STM Imaging with a Flexible Tip Apex. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2017**, 95 (4), 045407.
- (44) Ruffieux, P.; Cai, J.; Plumb, N. C.; Patthey, L.; Prezzi, D.; Ferretti, A.; Molinari, E.; Feng, X.; Müllen, K.; Pignedoli, C. A.; Fasel, R. Electronic Structure of Atomically Precise Graphene Nanoribbons. *ACS Nano* **2012**, *6* (8), 6930–6935.
- (45) Zheng, Y. J.; Huang, Y. L.; Chen, Y.; Zhao, W.; Eda, G.; Spataru, C. D.; Zhang, W.; Chang, Y.-H.; Li, L.-J.; Chi, D.; Quek, S.

- Y.; Wee, A. T. S. Heterointerface Screening Effects Between Organic Monolayers and Monolayer Transition Metal Dichalcogenides. *ACS Nano* **2016**, *10* (2), 2476–2484.
- (46) Liu, Z.-F.; Egger, D. A.; Refaely-Abramson, S.; Kronik, L.; Neaton, J. B. Energy Level Alignment at Molecule-Metal Interfaces From an Optimally Tuned Range-Separated Hybrid Functional. *J. Chem. Phys.* **2017**, *146* (9), 092326.
- (47) Thygesen, K. S.; Rubio, A. Renormalization of Molecular Quasiparticle Levels at Metal-Molecule Interfaces: Trends Across Binding Regimes. *Phys. Rev. Lett.* **2009**, *102* (4), 046802.